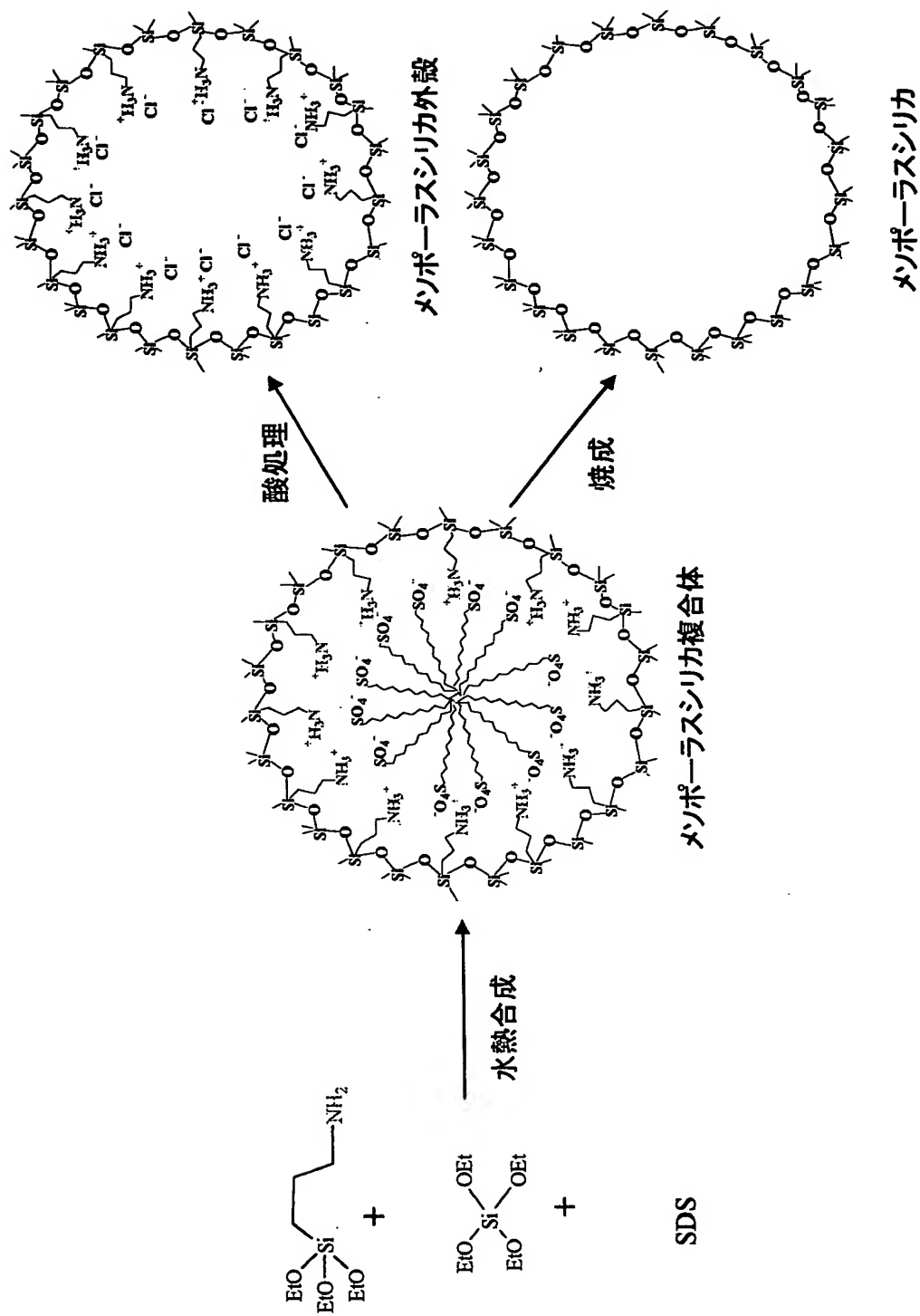
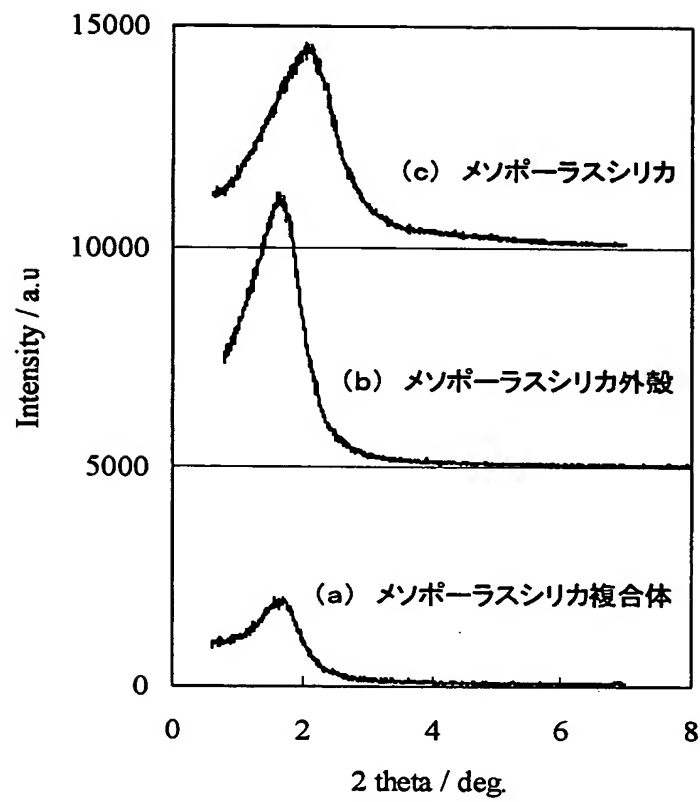


【図1】



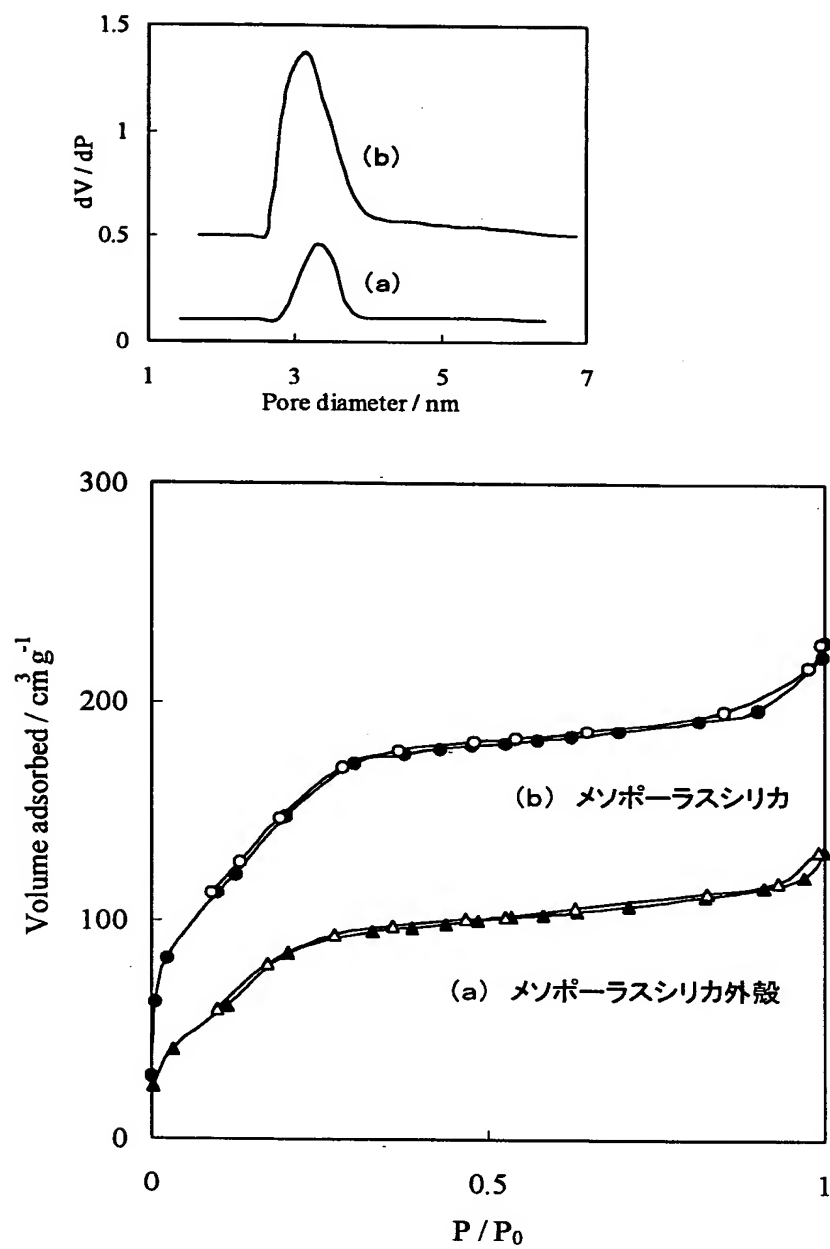
DF4392/US

【図 2】



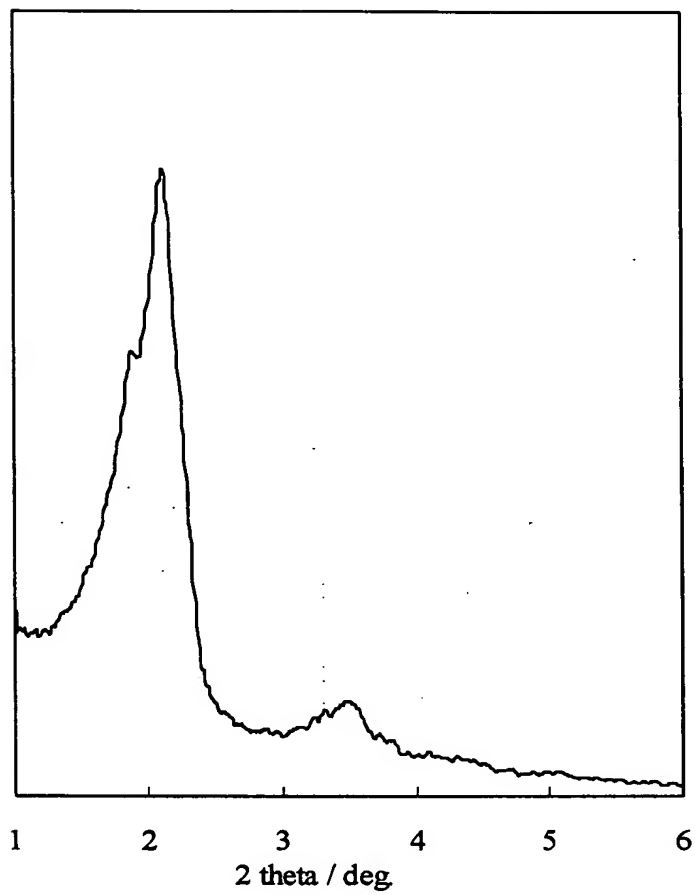
DF4392/US

【図 3】



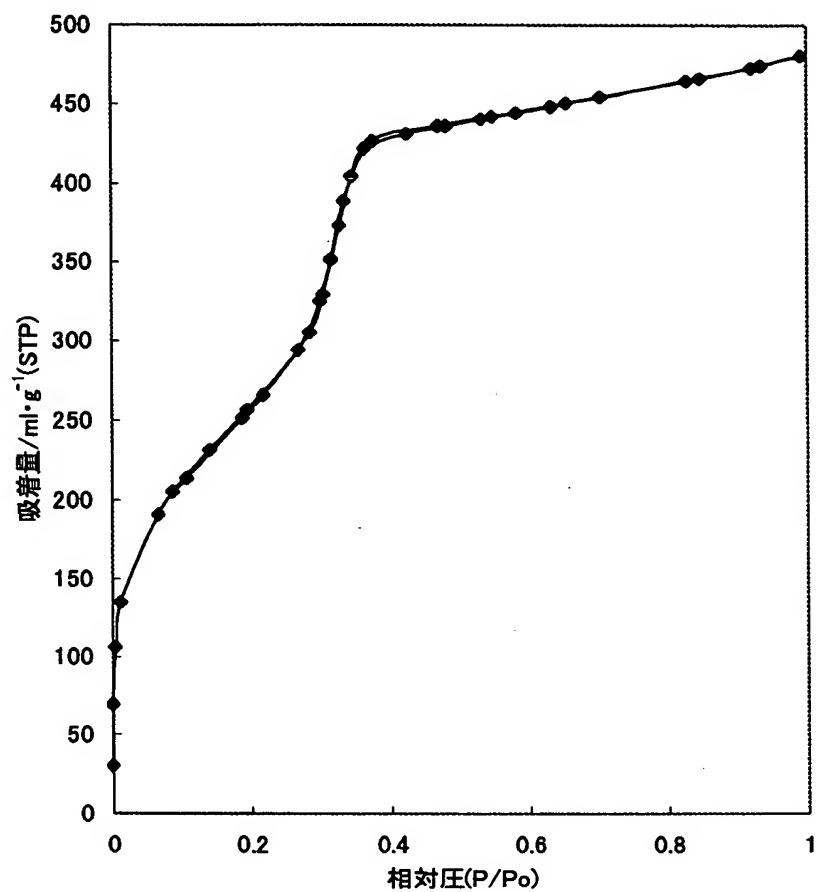
DF4392/US

【図 4】

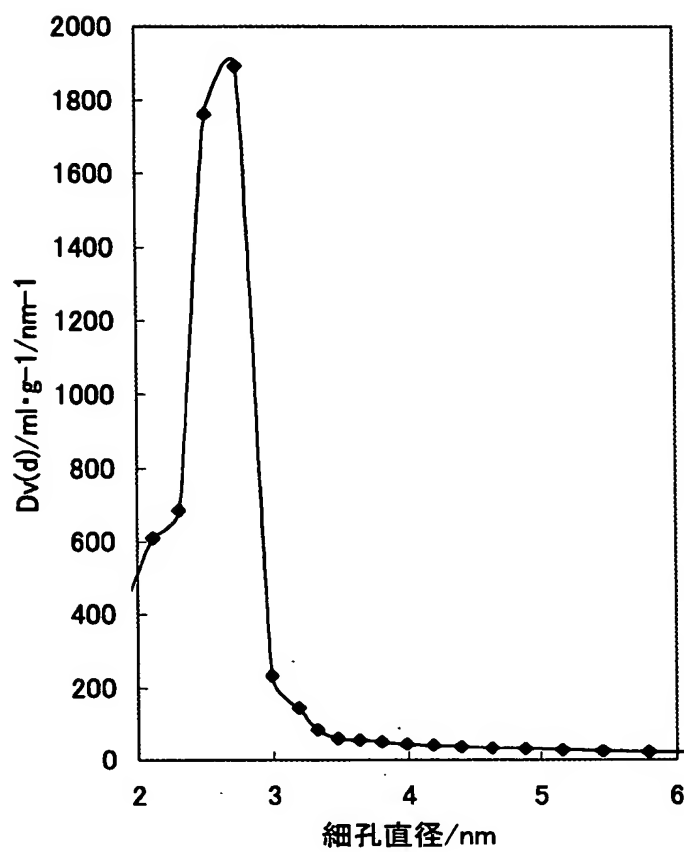


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【図 5】



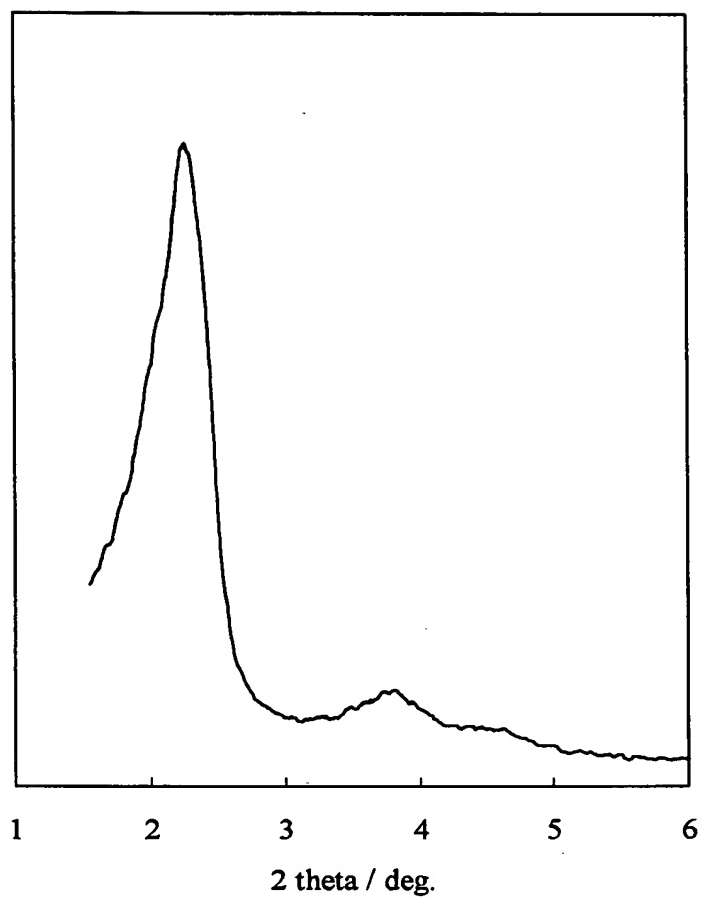
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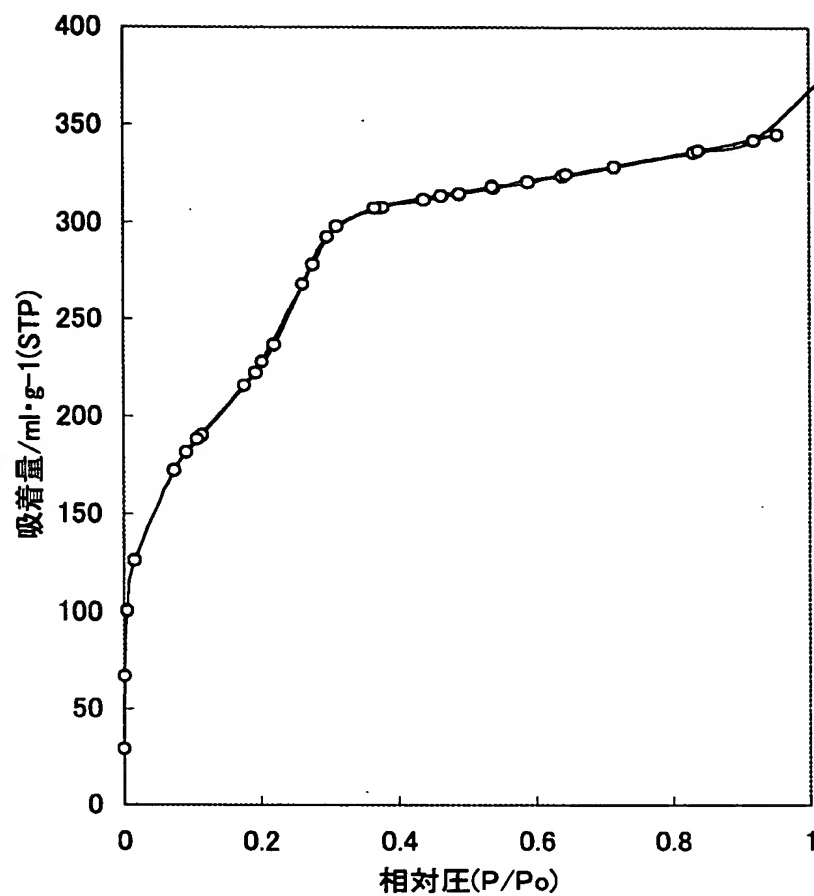
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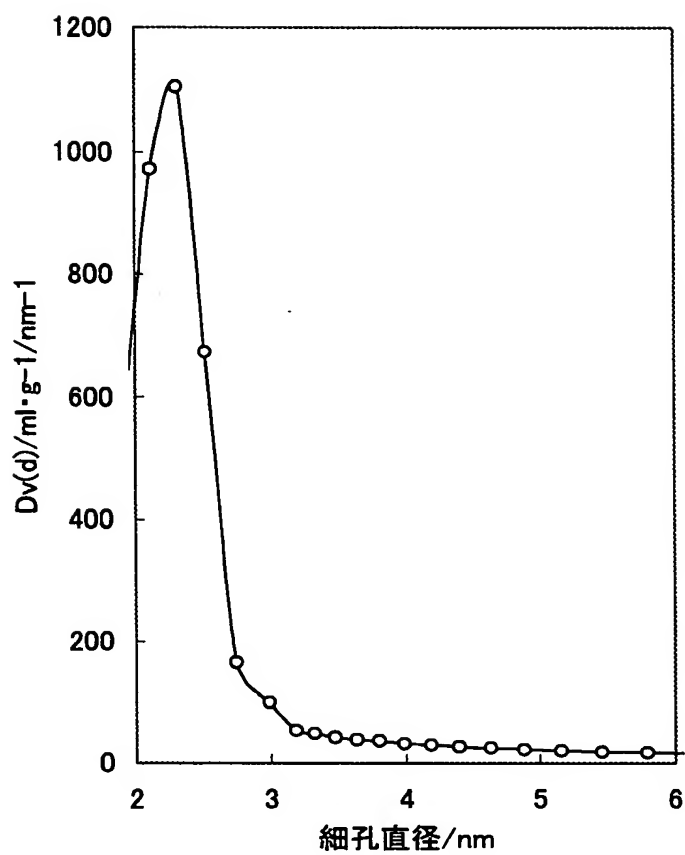
【図 7】



【図 8】



【図 9】



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	Neutralization	Double decomposition
Surfactant		
CSDA		
Interaction		
	<p>A: COO, OSO₃, SO₃, OPO₃; M⁺: Na⁺, K⁺, NH₄⁺ etc.; R₁: H, CH₃; n = 8 - 18;</p>	

Fig. 10. Schematic illustration of the two types of amino group-anionic surfactant head group interactions: through neutralization of acid with primary aminosilane APS and double decomposition of negatively charged anionic salt surfactant with positively charged quaternized aminosilane TMAPS.

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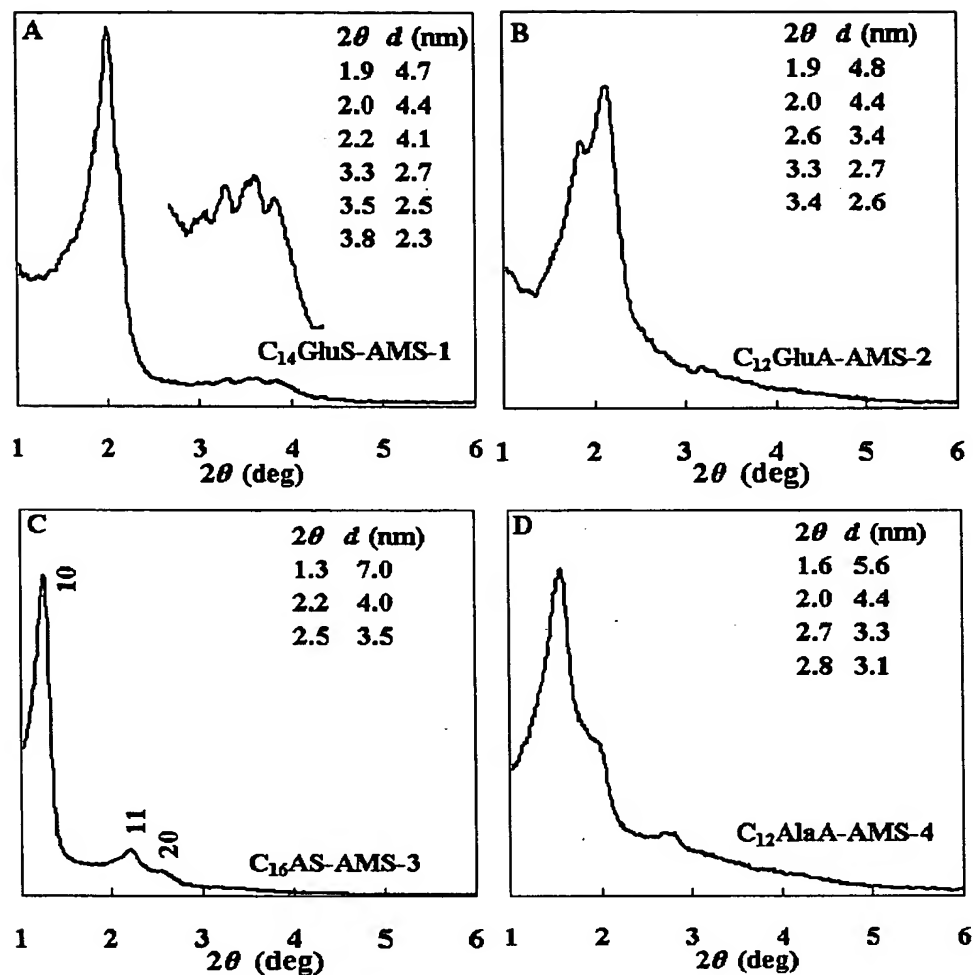


Fig. 11. XRD patterns of calcined AMS-n mesoporous silica. The chemical mol composition of the reaction mixture was (A) $C_{14}\text{GluS-AMS-1}$, $C_{14}\text{GluS:TMAPS:TEOS:H}_2\text{O}$ 1:2:10:2405 (at 100 °C for 3 d); (B) $C_{12}\text{GluA-AMS-2}$: $C_{12}\text{GluA:APS:TEOS:H}_2\text{O}$ 1:2.5:18.5:1905 (at 100 °C for 2 d); (C) $C_{16}\text{AS-AMS-3}$: $C_{16}\text{AS:TMAPS:TEOS:H}_2\text{O}$ 1:1:9:1544 (at 60 °C for 1 d); (D) $C_{12}\text{AlaA-AMS-4}$,

C₁₂AlaA:APS:TEOS:H₂O 1:0.75:7.5:1505 (at 60 °C for 1 d). XRD patterns were recorded on an MX Labo powder diffractometer equipped with Cu K α radiation (40 kV, 20 mA) at the rate of 1.0 deg/min over the range of 1.5 – 10.0 ° (2 θ).

Supporting online materials:

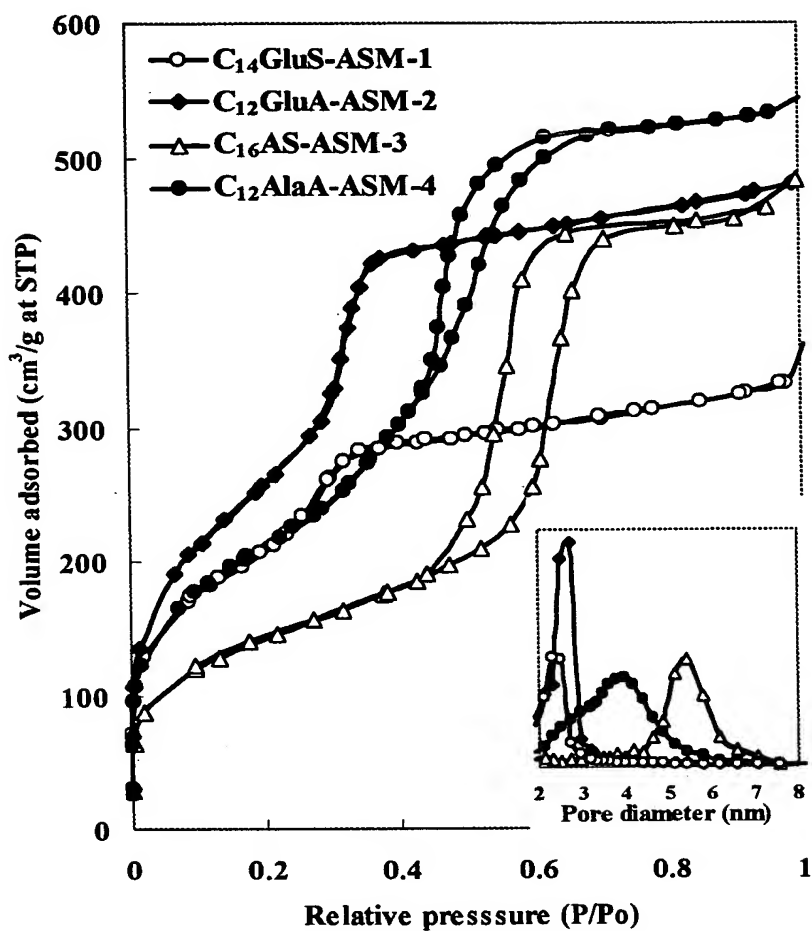


Fig. 12. N₂ adsorption-desorption isotherms and BJH pore size distributions of AMS-n mesoporous silica shown in Fig. 11. The isotherms were measured at -196 °C on a Belsorp 28SA sorptionmeter.

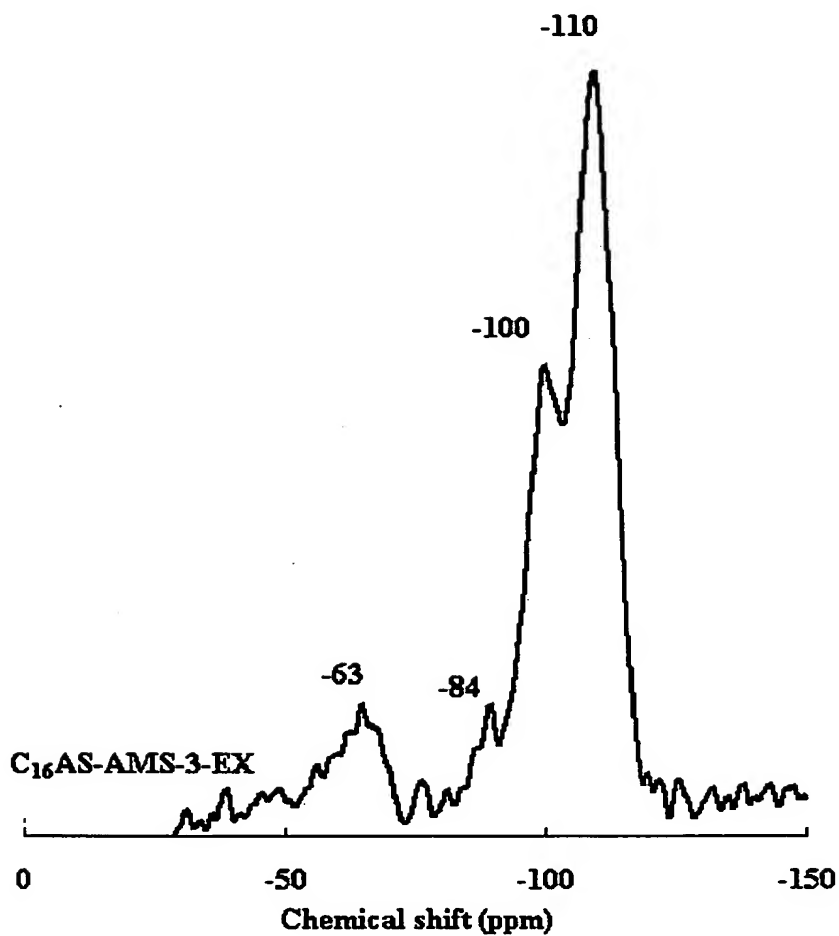


Fig. 13 shows CP ^{29}Si NMR spectra of extracted AMS-3 silica $\text{C}_{16}\text{AS-AMS-3-Ex}$. The spectra were collected at a JEOL-LA400WB 400 MHz spectrometer at 79.4 MHz and a sample spinning frequency of 5 kHz, respectively.